First-Principles Equation of State Database For Warm Dense Matter Computations



CODEC CENTER FOR MATTER UNDER EXTREME CONDITIONS

Center for Matter under Extreme Conditions (CMEC)







Outline

- Path integral Monte Carlo (PIMC) method
- Linear mixing approximation
- FPEOS database
- Comparison with different experiments

Path Integral Monte Carlo

Density functional molecular dynamics at lower T





Born-Oppenheimer approx. MD with classical nuclei:

F = m a

Forces derived DFT with electrons in the instantaneous ground state.

Path integral Monte Carlo at high T > 10⁴...10⁶ K







Starting from Restricted PIMC Simulations of Hydrogen

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Equation of State of the Hydrogen Plasma by Path Integral Monte Carlo Simulation

C. Pierleoni,^{1,2,*} D. M. Ceperley,³ B. Bernu,¹ and W. R. Magro³

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Molecular Dissociation in Hot. Dance Hydrogen									
Wolecular Dissociation in 110t, Dense Hyurogen									
N	W.R. Magro, ¹ D.M. Ceperley, ² C. Pierleoni, ³ and B. Bernu ⁴								

PIMC and **DFT-MD** Simulations of Hydrogen and Helium



Restricted PIMC for fermions: How is the restriction applied?



Free-particle nodes:

Construct a <u>fermionic trial density matrix</u> in form of a Slater determinant of single-particle density matrices:

$$\rho_T(R,R',\beta) = \begin{vmatrix} \rho(r_1,r_1',\beta) & \cdots & \rho(r_1,r_N',\beta) \\ \vdots & \ddots & \vdots \\ \rho(r_N,r_1',\beta) & \cdots & \rho(r_N,r_N',\beta) \end{vmatrix}$$

Enforce the following nodal condition for all time slices along the paths:

$$\rho_T[R(t), R(0), t] > 0$$

This 3N-dimensional conditions eliminates all negative and some positive contribution to the path \rightarrow Solves the fermion sign problem approx.

$$\rho_0^{[1]}(r,r';\beta) = \sum_k e^{-\beta E_k} \, \Psi_k(r) \, \Psi_k^*(r')$$

Silicates: MgSiO₃

MgSiO₃ : Principal Hugoniot Curve



Gonzalez, Soubiran, Peterson, Militzer, *Phys. Rev. B* **101** (2020) 024107

MgSiO₃ : Principal Hugoniot Curve



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Gonzalez, Soubiran, Peterson, Militzer, Phys. Rev. B 101 (2020) 024107



Inertial confinement fusion experiments with plastic coated spheres of liquid H₂



(Graphics: Bachmann et al. LLNL)

PIMC and DFT-MD simulations performed for C₂H, CH, C₂H₃, CH₃ and CH₄.













FPEOS database

Hugoniot Curves of BN and B₄C Fully interacting EOS and Linear Mixing agree quite well.



Hugoniot Curves of BN and B₄C Fully interacting EOS and Linear Mixing agree quite well.



Linear Mixing at Constant P and T (Also called additive volume rule)





$$V_{
m mix} = N_1 V_1 + N_2 V_2 ,$$

 $m_{
m mix} = N_1 m_1 + N_2 m_2 ,$
 $E_{
m mix} = N_1 E_1 + N_2 E_2 ,$
 $\rho_{
m mix} = m_{
m mix} / V_{
m mix}$

Hugoniot Curves of MgO and MgSiO₃ Results from fully interacting EOS and experiment.



Hugoniot Curves of MgO and MgSiO₃ Fully interacting EOS and Linear Mixing agree quite well.



Linear mixing works well for $T \gtrsim 2 \times 10^5$ K and $\varrho/\varrho_0 \gtrsim 3.2$

The Journal of Chemical Physics

Nonideal mixing effects in warm dense matter studied with first-principles computer simulations



Nonlinear Mixing Effects in MgSiO₃ Fully interacting EOS and Linear Mixing agree quite well.



Linear mixing works well for $T \gtrsim 2 \times 10^5$ K and $\varrho/\varrho_0 \gtrsim 3.2$

Hugoniot Curves of CO and CO₂

Experimental CO₂ Hugoniot agree with Linear Mixing result



Hugoniot Curves of H₂O, H₂O₂, and Al₂O₃ Experimental H₂O Hugoniot agree with Linear Mixing result





Material	Number	Minimum	Maximum	Minimum	Maximum	Number of	References
	of isochores	density	density	temperature	temperature	EOS points	
		$[\mathrm{gcm^{-3}}]$	$[\mathrm{gcm^{-3}}]$	[K]	[K]		
Hydrogen	33	0.001	798.913	15625	6.400×10^{7}	401	[69–74]
Helium	9	0.387	10.457	500	$2.048{ imes}10^9$	228	[75, 76]
Boron	16	0.247	49.303	2000	5.174×10^{8}	314	[77]
Carbon	9	0.100	25.832	5000	$1.035{ imes}10^9$	162	[78, 79]
Nitrogen	17	1.500	13.946	1000	$1.035{ imes}10^9$	234	[80]
Oxygen	6	2.486	100.019	10000	$1.035{ imes}10^9$	76	[81]
Neon	4	0.895	15.026	1000	$1.035{ imes}10^9$	67	[82]
Sodium	9	1.933	11.600	1000	$1.293{ imes}10^8$	193	[83, 84]
Magnesium	23	0.431	86.110	20000	5.174×10^{8}	371	[85]
Aluminum	15	0.270	32.383	10000	$2.156{ imes}10^8$	240	[86]
Silicon	7	2.329	18.632	50000	$1.293{ imes}10^8$	85	[87, 88]
LiF	8	2.082	15.701	10000	1.035×10^{9}	91	[89]
B_4C	16	0.251	50.174	2000	5.174×10^{8}	291	[90]
BN	16	0.226	45.161	2000	5.174×10^{8}	311	[91]
CH_4	16	0.072	14.376	6736	$1.293{ imes}10^8$	247	[92, 93]
CH_2	16	0.088	17.598	6736	$1.293{ imes}10^{8}$	248	[92, 93]
C_2H_3	16	0.097	19.389	6736	$1.293{ imes}10^{8}$	247	[92, 93]
CH	16	0.105	21.000	6736	$1.293{ imes}10^8$	248	[92, 93]
C_2H	16	0.112	22.430	6736	$1.293{ imes}10^{8}$	245	[92, 93]
MgO	19	0.357	71.397	20000	5.174×10^{8}	286	[94]
$MgSiO_3$	16	0.321	64.158	6736	5.174×10^{8}	284	[95, 96]

First-Principles Equation of State Database online http://militzer.berkeley.edu/FPEOS



5000 first-principles calculations have been combined into our FPEOS database. So anyone can predict shock Hugoniot curves for a variety of compounds and mixtures. This will make warm dense matter calculations more reliable and efficient.

NIF Gbar Experiment: Equations of State of C-O Mixtures in White Dwarf Stars

PI: D. Saumon (LANL), Blouin, Glenzer, Swift, Kritcher, Doppner, Whitley, Lazicki, Falcone, Militzer

We propose to make EOS measurements along the Hugoniot with the Gbar platform of carbon-oxygen rich materials that resemble conditions in White Dwarf stars.



Glyoxal C₂0₂H₂ comp:~/fpeos> fpeos binaryMixture EOS1=6 2.0 EOS2=18 2.0 rho0=1.27 E0=-227.8 Acetic acid C₂0₂H₄ comp:~/fpeos> fpeos binaryMixture EOS1=6 2.0 EOS2=16 2.0 rho0=1.049 E0=-229.0

FPEOS demo